

Review of previous lecture (April 21)

macroscopic observables in the microscopic ensemble

$$T^{-1} = \left. \frac{\partial S(E, V, N)}{\partial E} \right|_{V, N}$$

$$p = T \left. \frac{\partial S(E, V, N)}{\partial V} \right|_{E, N}$$

$$\mu = T \left. \frac{\partial S(E, V, N)}{\partial N} \right|_{E, V}$$

first law of thermodynamics:

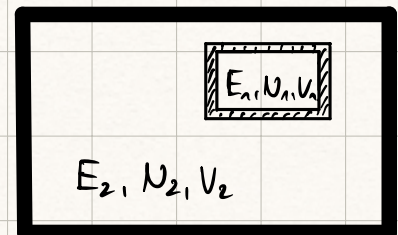
$$\frac{dE}{\text{state variables}} = T \frac{dS}{\text{state variables}} - p \frac{dV}{\text{state variables}} + \mu \frac{dN}{\text{state variables}}$$

Canonical ensemble:

$$E = E_1 + E_2, \quad E_2 \gg E_1$$

$$N = N_1 + N_2, \quad N_2 \gg N_1$$

$$V = V_1 + V_2, \quad V_2 \gg V_1$$



Hamiltonian: $H = H_1 + H_2 + W$

↳ energy exchange between subsystems

Consider expectation value of an operator A_n that only acts on states of system 1:

$$\begin{aligned}
 \langle A_n \rangle &= \text{Tr}(\rho A_n) \\
 &= \text{Tr}(\rho_{mc}^{(1+2)} A_n) \\
 &= \text{Tr}_1 \text{Tr}_2(\rho_{mc}^{(1+2)} A_n) \quad \rightarrow \text{integrated out states of system 2} \\
 &= \text{Tr}_1(\rho^{(1)} A_n)
 \end{aligned}$$

derive explicit form of $\rho^{(1)}$:

$$\begin{aligned}
 \rho^{(1)} &= \text{Tr}_2 \rho_{mc}^{(1+2)} \\
 &= \text{Tr}_2 \left(Z_{mc}^{(1+2)}(E) \right)^{-1} \underbrace{\sum_n |n\rangle\langle n|}_{\sum_{n_1, n_2} |n_1, n_2\rangle\langle n_1, n_2|} \underbrace{\delta_{H_1 + H_2, E}}_{\delta_{H_2, E - H_1}} \\
 &= \sum_{n_1} \frac{Z_{mc}^{(2)}(E - H_{n_1})}{Z_{mc}^{(1+2)}(E)} |n_1\rangle\langle n_1|
 \end{aligned}$$

here we can use the fact that $\frac{E_1}{E} \ll 1$ and expand:

$$Z_{mc}^{(2)}(E - H_{n_1}) = Z_{mc}^{(1)}(E) - \frac{\partial Z_{mc}^{(2)}}{\partial E} H_{n_1} + \dots$$

note that $Z_{mc}^{(2)}(E)$ is a rapidly varying function (typically $Z_{mc}(E) \sim E^N$)

\Rightarrow it is much more useful and efficient to expand the logarithm $\log(Z_{mc}^{(2)}(E - H_{n_1}))$ instead

$$\begin{aligned}
 \log Z_{mc}^{(2)}(E - H_{n_1}) &\approx \log Z_{mc}^{(2)}(E) - \frac{\partial \log Z_{mc}^{(2)}(E)}{\partial E} H_{n_1} \\
 &= \log Z_{mc}^{(2)}(E) - \frac{H_{n_1}}{k_B T}
 \end{aligned}$$

$$\Rightarrow Z_{mc}^{(2)}(E-H_n) \approx Z_{mc}^{(2)}(E) e^{-\frac{H_n}{k_B T}}$$

$$\rho^{(1)} = \frac{Z_{mc}^{(2)}(E)}{Z_{mc}^{(2)}(E)} \sum_n \underbrace{e^{-\frac{H_n}{k_B T}}}_{e^{-\frac{E_n}{k_B T}}} |\nu\rangle\langle\nu|$$

normalization constant

$$= c \cdot \text{Tr}_n e^{-\frac{H_n}{k_B T}}$$

partition function and density matrix of canonical ensemble

$$Z_c(T) = \text{Tr} \exp\left(-\frac{H_n}{k_B T}\right)$$

$$\rho_c = Z_c^{-1} \exp\left(-\frac{H_n}{k_B T}\right)$$

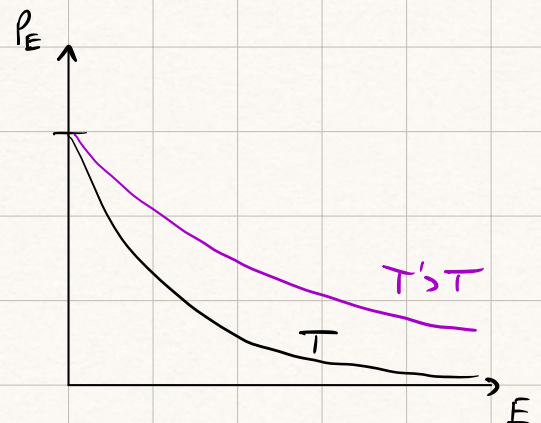
$$\text{Tr} \rho_c = 1$$

probability that system 1 has particular energy E_n

$$\rho^{(1)} = Z_c^{-1} \sum_n e^{-\frac{E_n}{k_B T}} |\nu\rangle\langle\nu|$$

$$= \sum_n p_{E_n} |\nu\rangle\langle\nu|$$

$$\hookrightarrow p_{E_n} = \frac{e^{-\frac{E_n}{k_B T}}}{Z_c}$$



entropy in canonical ensemble

$$\begin{aligned} S_c &= -k_B \cdot \text{Tr}(\rho_c \log \rho_c) \\ &= -k_B \text{Tr} \left[\rho_c \cdot \left(\log Z_c^{-1} - \frac{H}{k_B T} \right) \right] \\ &= +k_B \log Z_c \text{Tr} \rho_c + k_B \cdot \text{Tr} \left(\rho_c \frac{H}{k_B T} \right) \\ &= +k_B \log Z_c + \frac{1}{T} \underbrace{\text{Tr}(\rho_c H)} \\ &= +k_B \log Z_c + \frac{E}{T} \end{aligned}$$

↳ note that E is not fixed externally in the canonical ensemble, but given by an ensemble average of the Hamiltonian

note that entropy is not of the form $S = c \log Z_c$ anymore due to the additional term $\frac{E}{T}$

⇒ define a new state function of the canonical ensemble:

$$F = -k_B \cdot T \cdot \log Z_c(T) \quad \text{free energy}$$

$$\text{this implies: } S_c = -\frac{F}{T} + \frac{E}{T} \Rightarrow F = E - TS_c$$

the density matrix in the canonical ensemble depends on:

$$\rho_c = \rho_c(T, V)$$

↓ ↘ in Hamiltonian, e.g. via hard-wall interactions at boundaries
explicitly via $e^{-\beta H}$

$$\Rightarrow F = F(T, V) \text{ compared to } E = E(S, V)$$

$$dF = dE - T \cdot dS - S \cdot dT$$

$$= -p \cdot dV - S \cdot dT$$

$$(dU=0)$$

↳ Legendre transformation
(exercise)

thermodynamic observables in the canonical ensemble

$$E = \langle H \rangle = \text{Tr}(\rho_c H)$$

$$= z_c^{-1} \text{Tr}(H e^{-\beta H})$$

$$= -z_c^{-1} \frac{\partial}{\partial \beta} \text{Tr}(e^{-\beta H})$$

$$= -z_c^{-1} \frac{\partial}{\partial \beta} z_c = -\frac{\partial}{\partial \beta} \log z_c$$

$$P = -\langle \frac{\partial H}{\partial V} \rangle = -\text{Tr}(\rho_c \frac{\partial H}{\partial V})$$

$$= -z_c^{-1} \text{Tr}(\frac{\partial H}{\partial V} e^{-\beta H})$$

$$= +z_c^{-1} \beta^{-1} \frac{\partial}{\partial V} z_c$$

$$= +k_B T \frac{\partial}{\partial V} \log z_c$$

$$S_c = k_B \frac{\partial}{\partial T} (T \cdot \log z_c)$$

(using the relations above)

$$dF = d(-k_B T \log \text{Tr} e^{-\beta H})$$

$$= -k_B dT \underbrace{\log \text{Tr} e^{-\beta H}}_{\log z_c} - k_B \cdot T z_c^{-1} \text{Tr} \left[\left(\underbrace{\frac{dT}{k_B \cdot T^2} H}_{\frac{E}{k_B \cdot T^2}} - \frac{1}{k_B T} \frac{\partial H}{\partial V} \cdot dV \right) e^{-\beta H} \right]$$

$$= -\frac{dT}{T} (k_B T \log z_c + E) + \langle \frac{\partial H}{\partial V} \rangle dV$$

$$= -S_c \cdot dT - p \cdot dV \quad (+ p \cdot dN)$$

Various ensembles and its properties

ensemble	microcanonical	canonical	grandcanonical
System type	isolated	energy exchange with heatbath	energy + particle exchange
density matrix	$Z_{mc}^{-1} [\delta(E-H) - \delta(H-(E-\Delta E))]$	$Z_c^{-1} \exp(-\frac{H}{k_B T})$	$Z_{gc}^{-1} \exp(-\frac{H - \mu N}{k_B T})$
State variables	E, V, N	T, V, N	T, V, μ
state function	$S = +k_B \log Z_{mc}$	$F = -k_B T \log Z_c$	$\Phi = -k_B T \log Z_{gc}$

all thermodynamic quantities can be extracted from the partition function resp. the state function and its derivatives.

Key task in statistical physics is the computation of the partition function Z , typically Z_c (canonical ensemble) in this course.

Ergodicity

consider a classical system

a system is ergodic if

$$\lim_{T \rightarrow \infty} \frac{1}{T} \int_0^T dt A(\{x_i(t), p_i(t)\}) = \int \frac{dx^{3N} dp^{3N}}{h^{3N} N!} \underbrace{P_{eq}(\{x_i, p_i\})}_{\text{equilibrium probability distribution}} A(\{x_i, p_i\})$$

equilibrium probability
distribution

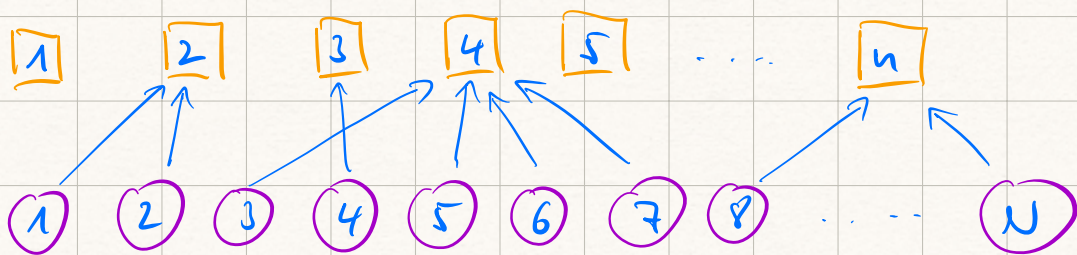
that means: time average = ensemble average

over sufficiently long periods of time the microstates of the system approaches every point in phase space

illustration: human behaviour and ergodicity

consider n restaurants ("phase space") and

N people who visit the restaurants on a regular basis



time average: pick one person and measure how often the different restaurants have been visited

ensemble average: snapshot of all people at a given time

Is this system ergodic?

generally not! Why? → find counter examples

How about systems in statistical physics?

also not, breaking of ergodicity closely related to phase transitions and spontaneous symmetry breaking

More soon!